

# Corrosion Glossary

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## natural aging

Spontaneous aging of a supersaturated solid solution at room temperature. See also *aging*. Compare with *artificial aging*.

## Nernst equation

An equation that expresses the exact *electromotive force* of a cell in terms of the activities of products and reactants of the cell.

## Nernst layer, Nernst thickness

The diffusion layer or the hypothetical thickness of this layer as given by the theory of Nernst, is defined by:

$$i_d = n F D (C^o - C)/d$$

where,  $i_d$  = the diffusion limited current density,  $D$  = the diffusion coefficient,  $C^o$  = the concentration at the electrode surface, and  $d$  = the Nernst thickness. It is a hypothetical thickness which has been found to be 0.05 cm in many cases of unstirred aqueous electrolytes.

## neutron embrittlement

*Embrittlement* resulting from bombardment with neutrons, usually encountered in metals that have been exposed to a neutron flux in the core of a reactor. In steels, neutron embrittlement is evidenced by a rise in the ductile-to-brittle transition temperature.

## nitriding

Introducing nitrogen into the surface layer of a solid ferrous alloy by holding at a suitable temperature (below  $A_{c1}$  for ferritic steels) in contact with a nitrogenous material, usually ammonia or molten cyanide of appropriate composition. Quenching is not required to produce a hard case.

## nitrocarburizing

Any of several processes in which both nitrogen and carbon are absorbed into the surface layers of a ferrous material at temperatures below the lower critical temperature and, by diffusion, create a concentration gradient. Nitrocarburizing is performed primarily to provide an anticreeping surface layer and to improve fatigue resistance. Compare with *carbonitriding*.

## noble

The positive direction of *electrode potential*, thus resembling noble metals such as gold and platinum.

## noble metal

(1) A metal whose *potential* is highly positive relative to the hydrogen electrode. (2) A metal with marked resistance to chemical reaction, particularly to oxidation and to evolution by inorganic acids. The term as often used is synonymous with *precious metal*.

## noble potential

A *potential* more cathodic (positive) than the standard hydrogen potential.

## normalizing

Heating a ferrous alloy to a suitable temperature above the transformation range and then cooling in air to a temperature substantially below the transformation range.

## normal solution

An aqueous solution containing one gram equivalent of the active reagent in 1L of the solution.

## normal stress

The stress component perpendicular to a plane on which forces act. Normal stress may be either tensile or compressive.